This article was downloaded by: On: 23 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713617200

Absolute Configuration of $4-\alpha$ -D-Glucopyranosyl-*myo*-inositol, Enzymatic Transglycosylation Product

Yutaka Watanabe^a; Chikara Nakamoto^a; Shoichiro Ozaki^a; Michikatsu Sato^b; Kyoko Koizumi^c ^a Department of Applied Chemistry, Faculty of Engineering, Ehime University, Matsuyama, Japan ^b Central Research Laboratories, Mercian Corporation, Fujisawa, Japan ^c Faculty of Pharmaceutical Sciences, Mukogawa Women's University, Nishinomiya, Japan

To cite this Article Watanabe, Yutaka , Nakamoto, Chikara , Ozaki, Shoichiro , Sato, Michikatsu and Koizumi, Kyoko(1993) 'Absolute Configuration of 4- α -D-Glucopyranosyl-*myo*-inositol, Enzymatic Transglycosylation Product', Journal of Carbohydrate Chemistry, 12: 6, 685 – 692

To link to this Article: DOI: 10.1080/07328309308018999 URL: http://dx.doi.org/10.1080/07328309308018999

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

J. CARBOHYDRATE CHEMISTRY, 12(6), 685-692 (1993)

ABSOLUTE CONFIGURATION OF $4-\alpha$ -D-GLUCOPYRANOSYL-*MYO*-INOSITOL, ENZYMATIC TRANSGLYCOSYLATION PRODUCT

Yutaka Watanabe,^a,* Chikara Nakamoto,^a Shoichiro Ozaki,^a Michikatsu Sato,^b and Kyoko Koizumi^c

 ^a Department of Applied Chemistry, Faculty of Engineering, Ehime University, Bunkyo-cho 3, Matsuyama 790, Japan
^b Central Research Laboratories, Mercian Corporation, 9-1, Johnan 4-chome, Fujisawa 251, Japan
^c Faculty of Pharmaceutical Sciences, Mukogawa Women's University, 11-68, Koshien Kyuban-cho, Nishinomiya 663, Japan

Received September 24, 1992 - Final Form February 10, 1993

ABSTRACT

The absolute configuration of $4-\alpha$ -D-glucopyranosyl-myo-inositol, one of the enzymatic transglycosylation products from cyclodextrin to myo-inositol, was confirmed by chemical synthesis. The glycosylation was shown to proceed in a highly diastereoselective manner by HPLC analysis of the chemically synthesized racemic glycoside.

INTRODUCTION

Recently we reported the synthesis of oligoglucosyl-inositols by transglucosylation of the cyclomaltodextrin glucanotransferase (CGTase from *Bacillus ohbensis*) using *myo*-inositol as the acceptor and β -cyclodextrin as the donor.¹ Structures of the products were

determined mainly by ¹H NMR spectroscopy. Among these glucosides, however, the absolute configuration of the inositol part in 4- α -D-glucopyranosyl-myo-inositol 1 could not be determined by NMR techniques since myo-inositol itself has a symmetry plane in which C-2 and C-5 are involved. We report here that the stereochemistry of the glycoside has been determined to be **1a**, and not **1b**, by an unambiguous chemical synthesis.



RESULTS AND DISCUSSION

We previously prepared optically active 1L- and 1D-1,3,4,5-tetra-O-benzoyl-myoinositol (L-2 and D-2) by kinetic resolution using methyl L- and D-2,3-Ocyclohexylidenetartrate respectively.² The former L-2 is a suitable starting material for derivatization to optically active nonabenzoate 6 which can be utilized for determination of the absolute configuration of 1a. Thus, L-2 with 74% ee $(L-2/D-2\ 87\ :\ 13)^3$ was selectively glucosylated at C-6 by reaction with 2,3,4,6-tetra-O-benzyl-D-glucopyranosyl fluoride⁴ (α/β 1:1) in the presence of trifluoromethanesulfonic anhydride⁵ to give the α -Dglucoside 4 in 76% yield together with β -D-glucoside (ca. 11% yield).⁶ Removal of the benzyl groups in 4 by hydrogenolysis on Pd-C followed by exhaustive benzoylation gave nonabenzoate 6 (66% yield), accompanied by the 2-hydroxy derivative 7. On the other hand, the enzymatic glycosylation product 1 1a was benzoylated at 60 °C to afford 6 in 96% yield while the reaction at room temperature gave 5-hydroxy derivative 8 in 86% yield as a major product along with 11% yield of 6. Interestingly, 7 was not formed in this case. The benzoate 6 thus derived from 1a showed an R_f value on silica gel TLC and spectra (IR and ¹H NMR) identical with those of chemically synthesized major diastereomer 6. The other diastereomer gave a different ¹H NMR spectrum by comparison with the NMR spectrum of a 1:1 diastereomeric mixture of DL-6 derived from racemic 1,3,4,5-tetrabenzoate DL-2. Consequently, the absolute configuration of the glycoside 1 has been unambiguously confirmed to be 1a.







Chart 1. HPLC Profiles of Enzyme-catalyzed (A) and Synthesized (B) Products. The last peak in (A) corresponds to $5-O-\alpha$ -D-glucopyranosyl-myo-inositol.

Evaluation of Diastereoselectivity of the Enzyme-Catalyzed Glycosylation. Deprotection of **DL-6** with sodium methoxide afforded 1 quantitatively. The HPLC analysis of 1 showed equal amounts of two peaks. The latter agreed with the larger peak of the enzyme-catalyzed glycosylation product 1a and the ratio of 1a and 1b was 98.8 : 1.2. Consequently, the enzymatic transglucosylation affording 1a has now been proved to proceed in a highly diastereoselective way.

Comparison of ¹³C NMR Spectra of 4-Glycosyl-myo-inositol Derivatives. The carbon-13 resonances for the inositol ring in 1a resembled quite closely those from $1L-4-O-\alpha-L$ -fucopyranosyl-myo-inositol⁷9a which is the product glycosylated at the enantiomericaly opposite C-4 position of inositol. Chemical shifts of C-3 and C-5 in **9a** were especially different from those in the corresponding diastereomer **9b** as shown in Table 1 and comparison of these shifts seems to be diagnostic for determining absolute configuration of 4-*O*-glycosyl *myo*-inositols. However, it should be noted that accordance of chemical shifts is not related to similarity of the absolute configuration since introduction of a chiral substituent on the hydroxyl group at C-4 in *myo*-inositol results in the formation of diastereomers. It is interesting to point out that there exists an enantiomer-like relation between **1a** and **9a** as illustrated in Scheme 2. This relation might be expressed by the use of the configurational symbols D and L. Thus, **1a** is expressed as D-D when the former and the latter symbols refer to the designations of *myo*-inositol and sugar moieties, respectively. Consequently, **9a** is shown as L-L. A similar enantiomeric relation was observed between 1D-4-O- α -L-fucopyranosyl-*myo*-inositol **9b**⁷ (= D-L) and 1L-4-O- α -D-glucopyranosyl-*myo*-inositol **1b** (= L-D) which was separated from **1a** by preparative HPLC of **1** (Table 1). Furthermore, **9b** (= D-L) showed chemical shifts of inositol carbons similar to those of 1L-4-O- α -D-mannopyranosyl-*myo*-inositol **10**⁸ (= L-D).



Scheme 2. Enantiomeric relations between 1a and 9a and between 9b and 10

Table 1.	¹³ C NMR Spectral	Data for	Inositol	Ring	Carbons in
	Compounds 1a	1h 9a	9h and	10	

	C-1	C-2	C-3	C-4	C-5	C-6		
1a	71.9	73.3	70.7	81.8	75.4	73.1		
9a	71.8	73.0	70.6	81.6	75.4	73.3 (Ref. 7)		
1b	71.8	73.0	72.4	81.7	73.7	73.4		
9b	71.8	73.0	72.5	81.3	73.7	73.3 (Ref. 7)		
10	72.0	73.5	72.6	80.8	73.9	73.6 (Ref. 8)		

EXPERIMENTAL

General Procedures. NMR spectra (¹H and ¹³C) were recorded on a JEOL GSX-270 and GSX-500. IR spectra were recorded on a Hitachi EPI-G3. Optical rotations were measured with a Union PM-101. Elemental analyses were performed with a Perkin-Elmer 240C. Thin layer chromatographic analyses were performed on Merck pre-coated plates, Silica Gel 60 F254. Anhydrous reaction atmosphere was achieved by nitrogen gas. Extracts obtained after work-up were dried over MgSO₄.

1L-1,3,4,5-Tetra-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzyl-α-D-glucopyranosyl)-myo-inositol (4). A solution of 2,3,4,6-tetra-O-benzyl-D-glucopyranosyl fluoride⁴ (3, α/β 1:1, 70 mg, 0.13 mmol) and 1L-1,3,4,5-tetra-O-benzoyl-myo-inositol (L-2, 74% ee, 38 mg, 0.063 mmol) in diethyl ether (2 mL) containing powdered 4A molecular sieves was cooled to -20 °C and trifluoromethane-sulfonic anhydride (10.6 µL, 0.063 mmol) was added. The mixture was stirred at room temperature for 18 h, filtered, and diluted with AcOEt. The organic solution was washed with water, aqueous NaHCO₃, and brine, dried, and concentrated. The oily residue was chromatographed on silica gelcoated glass plate (ethyl acetate-hexane, 2 : 5) to give 4 (54 mg, 76%) as an amorphous mass, R_f 0.29 (ethyl acetate-hexane, 2 : 5) and β-glycoside, R_f 0.38 (<7 mg, ca. 11%) which was contaminated with unidentified products.

4: mp 163-169.5 °C (MeOH and hexane: Diastereomeric ratio did not change before and after recrystallization); $[\alpha]_D{}^{21}$ 8.4° (*c* 1.4, CHCl₃), IR (CHCl₃ solution) 3650, 1730 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) δ 3.12 (m, 1H, *J* =10.7Hz), 3.25 (dd, 1H, *J* =9.8 and 3.4 Hz), 3.47 (m, 1H), 3.49 (t, 1H, *J* =9.8 Hz), 3.78 (m, 1H), 3.86 (t, 1H, *J* =9.8 Hz), 3.88-4.78 (complex, 8H), 4.63 (t, 1H, *J* =3.1 Hz), 4.68 (t, 1H, *J* =9.5 Hz), 4.88 (d, 1H, *J* =3.4 Hz, anomeric α -proton), 5.53 (dd, 1H, *J* =9.5 and 3.1 Hz), 5.54 (dd, 1H, *J* =9.5 and 3.1 Hz), 5.92 (t, 1H, *J* =9.5 Hz), 6.36 (t, 1H, *J* =9.5 Hz), 6.95-7.56 (complex, 32H), and 7.83-8.08 (complex, 8H); ¹³C (67.8 MHz) δ 67.71, 69.00, 69.87, 71.47, 71.99, 72.16, 72.47, 72.94, 73.21, 74.57, 75.38, 77.21, 77.57, 79.35, 81.34, 99.87, 127.25, 127.36, 127.45, 127.49, 127.77, 127.92, 128.00, 128.09, 128.16, 128.21, 128.24, 128.32, 128.43, 128.91, 128.99, 129.63, 129.77, 129.80, 129.93, 132.81, 133.06, 133.24, 133.41, 137.93, 138.44, 138.55, 138.81, 165.30, 165.40 (2C), and 165.82.

Anal. Calcd for C₆₈H₆₂O₁₅: C, 72.97; H, 5.68. Found: C, 72.59; H, 5.58.

1L-1,3,4,5-Tetra-O-benzoyl-6-O- α -D-glucopyranosyl-myo-inositol (5). Benzyl ether 4 in methanol and dichloromethane (1 mL each) was stirred with 5% Pd-C (212 mg) for 17 h at room temperature under a hydrogen atmosphere. Filtration and concentration to dryness gave tetrabenzoate **5** quantitatively (one spot on silica gel TLC); $R_{\rm f}$ 0.84 (MeOH-CH₂Cl₂ 1 : 2); IR (nujol) 3300, 1700 cm⁻¹; ¹H NMR (270 MHz, CD₃OD-CDCl₃ 1/5) δ 3.11-3.42 (complex, 6H), 4.62 (t, 1H, J = 2.4 Hz), 4.85 (t, 1H, J = 10.1 Hz), 5.15 (d, 1H, J = 3.7 Hz), 5.50 (dd, 1H, J = 10.1 and 2.4 Hz), 5.60 (dd, 1H, J = 10.1 Hz), 6.23 (t, 1H, J = 10.1 Hz), 7.20-7.65 (complex, 12 H), and 7.74-8.22 (complex, 8H).

1L-1,2,3,4,5-Penta-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzoyl- α -D-glucopyranosyl)-myo-inositol (6) and 1L-1,3,4,5-Tetra-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzoyl- α -D-glucopyranosyl)-myo-inositol (7). A mixture of 5 (35.1 mg, 0.046 mmol), benzoyl chloride (161 μ L, 1.388 mmol), and 4dimethylaminopyridine (a catalytic amount) in pyridine (2 mL) was stirred overnight at room temperature. After addition of H₂O, the mixture was stirred for an additional 10 min to decompose the excess of the chloride and diluted with AcOEt. The organic solution was washed with aqueous KHSO₄, aqueous NaHCO₃, and brine, dried, and concentrated. The residue was purified by preparative TLC (benzene-acetone, 10:1) to afford nonabenzoate **6** (39 mg, 66% yield) and octabenzoate **7** (12.5 mg, 23% yield).

6 (diastereomeric ratio, 77 : 23 as analyzed by ¹H NMR): $R_f 0.49$ (benzeneacetone, 12 : 1); mp 120-28 °C (from MeOH/hexane), IR (nujol) 1720 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) for the major diastereomer δ 3.67 (m, 1H), 4.34 (complex, 2H), 4.99 (t, 1H, J = 10.1 Hz), 5.21 (dd, 1H, J = 9.8 and 3.7 Hz), 5.57 (t, 1H, J = 9.8 Hz), 5.71 (d, 1H, J = 3.7 Hz, 1H), 5.79 (dd, 1H, J = 10.1 and 3.1 Hz), 5.91 (t, 1H, J = 9.8 Hz), 5.96 (dd, 1H, J = 10.1 and 3.1 Hz), 6.00 (t, 1H, J = 10.1 Hz), 6.22 (t, 1H, J = 10.1Hz), 6.27 (t, 1H, J = 3.1 Hz), and 7.16-8.19 (complex, 45 Hz).

Anal. Calcd for C₇₅H₅₈O₂₀: C, 70.42; H, 4.59. Found: C, 70.42; H, 4.61.

7: Rf 0.41 (benzene-acetone, 12 : 1); IR (CHCl₃) 3500, 1720 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) δ 3.99 (dd, 1H, J = 12.5 and 4.9 Hz), 4.42 (dd, 1H, J = 12.5 and 2.4 Hz), 4.52 (m, 1H), 4.72 (t, 1H, J = 2.4 Hz), 5.08 (t, 1H, J = 10.1 Hz), 5.15 (dd, 1H, J = 10.4 and 3.7 Hz), 5.46 (t, 1H, J = 10.4 Hz), 5.51 (dd, 1H, J = 10.1 and 2.4 Hz), 5.71 (d, 1H, J = 3.7 Hz), 5.75 (dd, 1H, J = 10.1 Hz), 5.90 (t, 1H, J = 10.4 Hz), 5.91 (t, 1H, J = 10.1 Hz), 6.20 (t, 1H, J = 10.1 Hz), 7.14-7.65 (complex, 24H), and 7.65-8.18 (complex, 16H).

Benzoylation of Enzymatic Glucosylation Product 1a. A pyridine (1 mL) solution of 1a [21 mg, 0.061 mmol, obtained by recrystallization from H₂O-EtOH for removal of a trace of 1b and subsequent preparative HPLC using CarbonexTM (vide infra)], benzoyl chloride (101.7 μ L, 0.88 mmol), and a catalytic amount of 4-

dimethylaminopyridine was stirred for 12 h at room temperature. After further addition of benzoyl chloride (407 μ L, 3.51 mmol) the mixture was stirred at 60 °C for 10 h. The work-up procedure and preparative TLC similar to the above mentioned benzoylation procedure yielded nonabenzoate 6 (75.5 mg, 96% yield).

6 (diastereomericaly pure): $[\alpha]_D^{31}$ +65.5° (c 1.05, CHCl₃); mp 141-2 °C (from benzene). The ¹H NMR spectrum of the compound 6 was identical with that of the predominant diastereomer 6 (the structure illustrated in the Scheme 1) derived chemically from tetrabenzoate L-2. The NMR data of the latter 6 are shown above.

When 1a (50 mg, 0.146 mmol) in pyridine (2 mL) was treated with benzoyl chloride (254.4 μ L, 2.192 mmol) and 4-dimethylaminopyridine (a catalytic amount) at room temperature for *ca.* 30 h, 1L-1,2,3,4-tetra-*O*-benzoyl-6-*O*-(2,3,4,6-tetra-*O*-benzoyl- α -D-glucopyranosyl)-*myo*-inositol (8) (148 mg, 86% yield) and 6 (21 mg, 11% yield) were isolated [preparative TCL (benzene-acetone 12 : 1)].

8: $R_f 0.41$ (benzene-acetone, 12 : 1); mp 180-1 °C (benzene); $[\alpha]_D^{31} + 68^\circ$ (c 1.02, CHCl₃), IR (nujol) 3500, 1720 cm⁻¹; ¹H NMR (270 MHz, CDCl₃) δ 3.75 (dd, 1H, J = 12.2 and 2.4 Hz), 4.12 (dd, 1H, J = 12.2 and 2.1 Hz), 4.13 (t, 1H, J = 10.1 Hz), 4.25 (m, 1H), 4.53 (t, 1H, J = 10.1 Hz), 5.35 (dd, 1H, J = 10.1 and 3.7 Hz), 5.66 (t, 1H, J = 10.1 Hz), 5.67 (dd, 1H, J = 10.1 and 2.8 Hz), 5.74 (dd, 1H, J = 10.1 Hz), 5.87 (t, 1H, J = 10.1 Hz), 5.92 (d, 1H, J = 3.7 Hz), 6.07 (t, 1H, J = 10.1 Hz), 6.17 (t, 1H, J = 2.8 Hz), 7.20-7.57 (complex, 24H), and 7.64-8.10 (complex, 16H).

Anal. Calcd for C₆₈H₅₄O₁₉: C, 69.50; H, 4.63. Found: C, 69.24; H, 4.72.

DL-4-O-a-D-Glucopyranosyl-myo-inositol (1) from Nonabenzoate DL-6 and Evaluation of Diastereoselectivity of the Glycosylation. To a suspension of benzoate **DL-6** (derived from racemic tetrabenzoate **DL-2**) in MeOH (5 mL) was added 60% NaH in mineral oil (18.6 mg, 0.465 mmol) and the mixture was stirred at room temperature overnight and then NaH (18.6 mg, 0.465 mmol) was further added. The solution was stirred for 2 days and neutralized by addition of cation exchange resin [Diaion SK1B (Mitsubishi Chemical Industries Ltd.), H⁺-form]. The resin was filtered and washed with distilled water. The filtrate was passed through anion exchange resin (Diaion SA11A, OH⁻- form) and washed with distilled water, and concentrated. The residue was washed with hexane, dichloromethane, and ethyl acetate, and dried under reduced pressure to afford 1 (14.5 mg, quant.); ¹H NMR (270 MHz, CDCl₃) δ 3.10-3.94 (complex, 12H), 5.11 (d, 1H, J = 3.7 Hz), and 5.16 (d, 1H, J = 3.7 Hz). HPLC analysis of 1 and 1a was carried out by the use of Hypercarb[™] (Shandon Ltd., 100 x 4.7 mm i.d., other conditions; eluent: CH3CN-H2O 1 : 99, flow rate: 0.8 mL/min, column temperature: 40 °C). The ratio of 1a and 1b in the fraction of mono-glucosylated myoinositols obtained by the enzyme-catalyzed glucosylation was determined to be 98.8: 1.2.

Separation of DL-4-O- α -D-Glucopyranosyl-myo-inositol (1) to 1a and 1b by Preparative HPLC. Diastereomeric mixture of 1 thus obtained in the preceding experiment was separated by HPLC on a graphitized carbon column [Column: CarbonexTM (Tonen Ltd., 150 x 10 mm i.d.), eluent: CH₃CN-H₂O 3 : 97, flow rate: 0.7 mL/min, column temperature: ambient] to give 1a and 1b.

1b: $[\alpha]_D^{24} + 55.6^\circ$ (*c* 0.45, H₂O); ¹H NMR (270 MHz, D₂O, HOD as the reference=4.64 ppm) δ 3.22-3.69 (complex, 10H), 3.89 (dt, 1H, *J* =10.36 and 3.59 Hz), 3.92 (t, 1H, *J* =2.75 Hz), 5.15 (d, 1H, *J* =3.97 Hz); ¹³C NMR (125.65 MHz) δ 61.26, 70.27, 71.81, 72.35, 72.59, 72.77, 73.00, 73.39, 73.74, 73.89, 81.75, and 100.28.

REFERENCES AND NOTES

- M. Sato, K. Nakamura, H. Nagano, Y. Yagi, and K. Koizumi, *Biotechnology Lett.*, 14, 659 (1992); M. Sato, T. Matsuo, N. Orita, and Y. Yagi, *ibid.*, 13, 69 (1991).
- 2. Y. Watanabe, A. Oka, Y. Shimizu, and S. Ozaki, *Tetrahedron Lett.*, **31**, 2613 (1990).
- The absolute configuration of tetrabenzoate L-2 was confirmed by the comparison with the 1L-1,3,4,5-tetra-O-benzoyl-myo-inositol (L-2) derived from known 1D-2,4di-O-benzyl-myo-inositol. [S. Ozaki, Y. Kondo, H. Nakahira, S. Yamaoka, and Y. Watanabe, *Tetrahedron Lett.*, 28, 4691 (1987); G. Baudin, B. I. Glänzer, K. S. Swaminathan, and A. Vasella, *Helv. Chim. Acta*, 71, 1367 (1988); C. E. Dreef, R. J. Tuinman, C. J. J. Elie, G. A. van der Marel, and J. H. van Boom, *Recl. Trav. Chim. Pays-Bas*, 107, 395 (1988).] This result will be described elsewhere.
- 4. T. Mukaiyama, Y. Hashimoto, and S. Shoda, Chem. Lett., 935 (1983).
- H. P. Wessel, *Tetrahedron Lett.*, **31**, 6863 (1990); H. P. Wessel, G. Englert, and P. Stangier, *Helv. Chim. Acta*, **74**, 683 (1991).
- β-Glycoside contained a small amount of unidentified products and was not purified completely.
- 7. P. J. Garegg, B. Lindberg, I. Kvarnström, and S. C. T. Svensson, *Carbohydr. Res.*, 139, 209 (1985).
- C. J. J. Elie, R. Verduyn, C. E. Dreef, D. M. Brounts, G. A. van der Marel, and J. H. van Boom, *Tetrahedron*, 46, 8243 (1990).